

IMPACTS AND APPLICATIONS OF HYDROGEN INJECTION IN BLAST FURNACES: A REVIEW

IMPACTOS E APLICAÇÕES DA INJEÇÃO DE HIDROGÊNIO EM ALTOS-
FORNOS: UMA REVISÃO

IMPACTOS Y APLICACIONES DE LA INYECCIÓN DE HIDRÓGENO EN ALTOS
HORNOS: UNA REVISIÓN

David Hebert Vieira¹, Luciana Gomes Castanheira², Paulo Santos Assis³

DOI: 10.54899/dcs.v22i81.3303

Recibido: 02/09/2025 | Aceptado: 04/09/2025 | Publicación en línea: 11/09/2025.

ABSTRACT

The steel industry is one of the largest contributors to global CO₂ emissions, and the transition to low-carbon production methods is crucial to achieving climate goals. Hydrogen injection into blast furnaces (BFs) has emerged as a promising technology to reduce carbon emissions and improve the sustainability of steel production, especially when the hydrogen comes from non-fossil sources. This review comprehensively examines the impacts and applications of hydrogen injection in BFs, focusing on thermodynamic, operational and environmental aspects. By analyzing recent studies, it highlights the benefits, challenges and future directions by producing hot metal from hydrogen. The review synthesizes findings from more than 90 studies, providing an understanding of how hydrogen injection (HI) influences operations. Some parameters were emphasized, such as theoretical flame temperature (RAFT), oxygen enrichment (OE), coke consumption (CR), pulverized coal injection rate (PCI), blowing temperature (TS), top gas temperature (TGT) and CO₂ emissions. The potential of hydrogen to replace traditional carbon-based reducers was discussed. Overall, HI can reduce coke consumption and greenhouse gas emissions, increase process productivity, but will require technical care to meet expectations.

Keywords: CO₂ Emissions. Hydrogen. Injection. Blast Furnace.

RESUMO

A indústria siderúrgica é uma das maiores contribuintes para as emissões globais de CO₂, e a transição para métodos de produção com baixas emissões de carbono é crucial para alcançar as metas climáticas. A injeção de hidrogênio em altos-fornos (BFs) surgiu como uma tecnologia promissora para reduzir as emissões de carbono e melhorar a sustentabilidade da produção de aço, especialmente quando o hidrogênio provém de fontes não fósseis. Esta revisão examina de

¹ PhD Student in Materials Engineering, Universidade Federal de Ouro Preto (UFOP), Instituto Federal de Minas Gerais (IFMG), Ouro Preto, Minas Gerais, Brazil. E-mail: david.vieira@ifmg.edu.br

Orcid: <https://orcid.org/0009-0005-7692-1820>

² PhD in Materials Engineering, Universidade Federal de Ouro Preto (UFOP), Ouro Preto, Minas Gerais, Brazil. E-mail: luciana.castanheira@ufop.edu.br

³ PhD in Metallurgy, Universidade Federal de Ouro Preto (UFOP), Ouro Preto, Minas Gerais, Brazil. E-mail: Assis@ufop.edu.br Orcid: <https://orcid.org/0000-0003-0874-4162>

forma abrangente os impactos e as aplicações da injeção de hidrogênio em BFs, com foco nos aspectos termodinâmicos, operacionais e ambientais. Ao analisar estudos recentes, ela destaca os benefícios, desafios e direções futuras da produção de ferro gusa a partir do hidrogênio. A revisão sintetiza as conclusões de mais de 90 estudos, proporcionando uma compreensão de como a injeção de hidrogênio (HI) influencia as operações. Alguns parâmetros foram enfatizados, como temperatura teórica da chama (RAFT), enriquecimento de oxigênio (OE), consumo de coque (CR), taxa de injeção de carvão pulverizado (PCI), temperatura de sopro (TS), temperatura do gás de topo (TGT) e emissões de CO₂. O potencial do hidrogênio para substituir os redutores tradicionais à base de carbono foi discutido. No geral, a HI pode reduzir o consumo de coque e as emissões de gases de efeito estufa, aumentar a produtividade do processo, mas exigirá cuidados técnicos para atender às expectativas.

Palavras-chave: Emissões de CO₂. Injeção. Hidrogênio. Alto-Forno.

RESUMEN

La industria siderúrgica es una de las que más contribuye a las emisiones globales de CO₂, y la transición a métodos de producción con bajas emisiones de carbono es fundamental para alcanzar los objetivos climáticos. La inyección de hidrógeno en altos hornos se ha convertido en una tecnología prometedora para reducir las emisiones de carbono y mejorar la sostenibilidad de la producción de acero, especialmente cuando el hidrógeno procede de fuentes no fósiles. Esta revisión examina de forma exhaustiva los impactos y las aplicaciones de la inyección de hidrógeno en altos hornos, centrándose en los aspectos termodinámicos, operativos y medioambientales. Mediante el análisis de estudios recientes, destaca las ventajas, los retos y las perspectivas de futuro de la producción de arrabio a partir de hidrógeno. La revisión sintetiza los resultados de más de 90 estudios, lo que permite comprender cómo influye la inyección de hidrógeno (HI) en las operaciones. Se hicieron hincapié en algunos parámetros, como la temperatura teórica de la llama (RAFT), el enriquecimiento de oxígeno (OE), el consumo de coque (CR), la tasa de inyección de carbón pulverizado (PCI), la temperatura de soplado (TS), la temperatura del gas superior (TGT) y las emisiones de CO₂. Se discutió el potencial del hidrógeno para sustituir a los reductores tradicionales basados en el carbono. En general, la HI puede reducir el consumo de coque y las emisiones de gases de efecto invernadero, así como aumentar la productividad del proceso, pero requerirá cuidados técnicos para cumplir con las expectativas.

Palabras clave: Emisiones de CO₂. Inyección. Hidrógeno. Alto horno.



Esta obra está bajo una [Licencia Creative Commons Atribución- NoComercial 4.0 Internacional](https://creativecommons.org/licenses/by-nc/4.0/)

INTRODUCTION

The iron and steel sector is the largest CO₂ emitter among heavy industries and the second largest energy consumer, releasing 2.6–2.8 Gt annually (1–3). Its strong dependence on coal, responsible for 75% of energy demand and coke production, reinforces the role of blast furnaces

(BFs) for their efficiency and scale (4,5). Emission intensity varies among countries according to the share of electric arc furnaces (6).

In line with climate goals, low-carbon alternatives have gained prominence, such as biomass (7,8), hydrogen (9) and renewable electricity (10). Co-injection of H₂, O₂ and CO₂ with pulverized coal can raise productivity and reduce emissions (11). Using reactive coke with hydrogen has also proven effective (12). Hydrogen improves the reduction kinetics of iron oxides (13) and can partly replace coke, lowering carbon use and increasing efficiency (14–16), making it one of the main decarbonization paths (17). In China, hydrogen metallurgy could cut emissions by up to 62.53% by 2050 (18). Although natural gas injection is established, direct use of H₂ as a reductant requires further study, since its physical properties demand operational adaptations (19).

H₂ injection may reduce BF emissions by 16% or more (20), but faces challenges such as high renewable hydrogen costs (21,22) and thermal impacts (23). Trials have shown success with H₂-rich gas in different furnace zones (24,25) and with alternative carbonaceous injectants such as biomass residues (26–28). Large projects such as COURSE50 (Japan), ULCORED and HYBRIT (Europe) and U.S. initiatives explore H₂-based routes. Since 2010, Chinese efforts have reported significant reductions in fuel use and CO₂ (29,30).

Hydrogen affects flame temperature, slag composition, steel quality and coke performance (31), requiring adjustments such as oxygen enrichment (32). This review analyzes the operational and environmental impacts of H₂ injection in BFs, highlighting trends, challenges and perspectives for lower-carbon ironmaking.

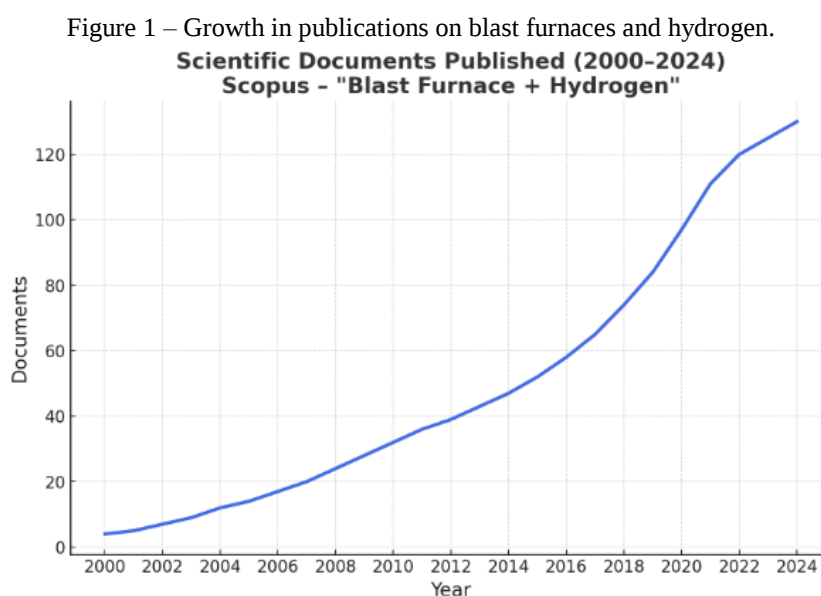
METHODOLOGY

This study addresses gaps in knowledge on hydrogen injection in blast furnaces. A literature review was carried out in October 2024 with the following steps:

- Initial Scopus search (“blast furnace” + “hydrogen injection”) found 35 documents (1984–2024), 26 post-2020, considered insufficient.
- A broader Scopus search (“blast furnace” + “hydrogen”) returned 1031 results, 476 (46.2%) from 2020–2024.
- From these, 50 relevant works were filtered, totaling 76.
- Web of Science search added 22, of which 20 overlapped.

- Connected Papers portal was used to expand related references.
- Sectoral statistics were also consulted.

In total, 90 documents were analyzed, extracting gaps, objectives, methods, results and conclusions. Data were organized in a spreadsheet (authors, year, title, journal, database). Around 26% of studies used CFD (1D–3D), 21% laboratory experiments, 20% mathematical models, 14% reviews, 7% thermodynamic models, and the rest various simulations and BF trials. Figure 1 shows the growth in publications on “blast furnace” and “hydrogen” in Scopus (2000–2024).



Source: Prepared by the authors themselves.

THEORETICAL FRAMEWORK

I. Thermodynamic effects

Reduction of iron oxides by H_2 is endothermic, while by CO it is exothermic. For instance: $Fe_2O_3 + 3H_2 \rightarrow 2Fe + 3H_2O$, $\Delta H = +100$ kJ (298 K); $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$, $\Delta H = -23.5$ kJ (298 K); $C + H_2O \rightarrow CO + H_2$ is also endothermic (33).

Thus, H_2 alters BF thermodynamics. It reduces oxides at lower temperatures than CO , lowering the cohesive zone (34–37). Reduction of FeO by H_2 is efficient below 1250 °C (up to 25% heat supplied by H_2), but excessive injection may cause incomplete reduction and higher coke use (4). Gas composition shifts, with more H_2 in the shaft, enhancing indirect reduction (16,38), though thermal stability in upper zones becomes harder to maintain.

Main effects: cooler upper furnace, thinner cohesive zone, higher ore/coke ratio, increased productivity, lower pressure loss and coke rate, and reduced emissions (14). Direct reduction

decreases with higher H_2 , while indirect reduction by H_2 rises, replacing part of CO reduction (16,39,40).

H_2 shifts the reserve zone equilibrium ($FeO + CO \leftrightarrow Fe + CO_2$) to lower temperatures. Its injection increases heat demand at the tuyere, usually compensated by higher O_2 or hotter blast. Above 1000 °C, H_2 mainly acts as a catalyst; below, it absorbs heat and affects the granular zone (23). It also shifts $H_2O + CO \leftrightarrow H_2 + CO_2$ equilibria. At high injection, gasification by H_2O raises CO in BF gas (14). Classic Chaudron diagrams show that below ~800 °C H_2 requires more moles than CO for full reduction, the reverse occurring at higher temperatures (41).

II. Effects on the Theoretical Flame Temperature (RAFT)

The theoretical (adiabatic) flame temperature (RAFT) is the temperature of gases leaving the combustion zone (CZ), influenced by fuel type, injection rate, blast conditions and material composition. With hydrogen, RAFT decreases due to its low inlet temperature and stronger cooling effect ($\approx 5\times$ that of natural gas, NG). In co-injection, H_2 lowers RAFT by ~ -4.7 °C/kg; alone, the effect is -14 °C/kg for H_2 and -2.8 °C/kg for NG (38). A way of calculating the RAFT, considering hydrogen injection, is shown in equation 1.

$$\text{Eq. 1} \quad \text{RAFT} = 2239.643 - 15I_H + 32.192F_0 - 0.629W_c - 3248,12BH.$$

where

I_H is the hydrogen injection in kg/thm, F_0 is the blowing enrichment rate in %, W_c is the PCI and BH is the blowing air humidity in m^3/m^3 . (42)

Cold H_2 requires higher O_2 enrichment, significantly reducing RAFT. For instance, 250 m^3/thm at 25 °C gave RAFT ≈ 1900 °C; with 350 m^3/thm , enrichment rose from 3.76% to 24.56%. Preheating H_2 (450–1250 °C) reduces O_2 demand and coke rate, expanding the operating window: at RAFT 1900 °C and TGT 130 °C, max injection rose from 237 m^3/thm (25 °C) to 350 m^3/thm (1250 °C) (32). With 120 m^3/thm H_2 , RAFT fell from 2171.83 °C to 1900 °C, requiring 10.38% O_2 enrichment (14). Table 01 summarizes published RAFT values.

Greater O_2 enrichment enables higher H_2 injection, though with declining efficiency. Raising blast temperature also allows higher rates for the same O_2 level. However, above 29–30% O_2 , Δt_{min} in the thermal reserve zone drops below 20 °C, impairing heat exchange and stability (43). Preheating H_2 reduces coke rate but not always CO_2 , since heating consumes

energy. The process of injecting coal together with H₂ preheated to 900 °C can increase the emissions reduction effect by around 30% with the use of green hydrogen (44).

Compared with other H₂-rich gases, pure H₂ lowers RAFT most and raises TGT most, while COG and shale gas give more stable TGT. At 1100 °C blast and 25 °C preheating, max injection rates were 90 kg/thm (shale gas), 107.2 (COG) and 31.8 (H₂). At 1200 °C blast, they rose to 90.4, 108.3 and 32.9; with 600 °C preheating, to 94, 112.5 and 32.1 kg/thm (45). Injecting cold H₂ increases TGT, while reducing the coke/ore ratio and increasing O₂ reduces TGT. Preheating the H₂ widens the operating window, allowing higher injection rates without compromising RAFT and TGT (32).

Table 1 - RAFT

RAFT °C	EO ₂ (%)	HI (m ³ /t)	HI (kg/t)	T Blow °C	PCI (kg/t)	Reference
2094	9	159	14,29	1273	120	(16)
2091	16,8	277	24,90	1273		
2098	15	304	27,33	1273		
2097	18,5	359	32,27	1273		
2040	12,4	160	14,38	1052	137	(46)
2051	4,03	217	19,54	1153	102,6	(36)
2051	4,03	217	19,54	1153		(47)
2051	3,65	209	18,82	1153		
2050	6,18	211	18,98	1153	104,1	(20)
2051	0,49	159	14,29	1153	140	
2000	14	300	26,97	1041	167	(20)

Source: Prepared by the authors themselves

RAFT decreases ~14.2 °C per kg of H₂ injected, reaching 1910 °C at 35 kg/thm (38). With $\Delta T_{min} = 20$ °C, RAFT ranged 2133–2266 °C with 0–12% enrichment at 1200 °C blast (35). Lower RAFT supports greater coke replacement and CO₂ reduction (48). Tuyere injection of reducing gas kept RAFT constant at 2314 °C despite higher O₂, preventing overheating in the lower furnace (49). For NG, preheating +100 °C increased RAFT by 17 °C, reduced TGT by 4.5 °C, and lowered coke rate by 7 kg/thm at 873 °C (50). The injection of hydrogenated gases is promising for reducing CO₂ and coke consumption. TGT is a limiting factor, with coke oven gas and pure H₂ being the most effective in reducing emissions (51). In general, TGT decreases with increasing H₂ content in the gas (5).

III. Fuel consumption

Hydrogen injection (HI) has strong potential to lower coke rate (CR). Reported results vary with injection mode (tuyere, shaft, co-injection). Comparative findings:

- Pure H₂ reduces CR more than COG or mixtures (16).

- Mass/heat balance simulations: max injection 19.5 kg/thm (RAFT/TGT limited), saving 37.5 kg/thm PCI (52).
- Increasing H₂ lowers chemical carbon use, but combustion consumption in CZ rises to supply heat (53).
- Tuyere injection reduces fuel rate at high O₂ (>35%), achieving lower values than BFs without injection (49).
- Higher H₂ volumes cut coke rate, while greater H₂ purity at constant volume may raise it (54).
- Coke rate decreases sharply with H₂ (33).
- Optimal CR reduction occurs at ~33.3% H₂; above this, CR rises again (40).
- One model estimated minimum coke consumption at 426 kg/thm under certain conditions. Reducing the degree of direct reduction from 27% to 21% would reduce consumption by 24 kg/thm (55).

Table 2 - Effect of Hydrogen Injection on Fuel Consumption

Hydrogen injection	Reduction in consumption	Reference
HI of 50 m ³ /thm	8.66% in CR.	(15)
HI between 7.9 and 10.9 m ³ /thm	Reduction of 1 kg/thm of CR, depending on the injection position (tuyere or shaft).	(43)
1 kg of H ₂	Reduction of between 3 and 4 kg	(35)
12.2 kg/t of H ₂	8% CR	(2)
For every 50 m ³ /thm of H ₂ at 25 °C added, With the injection of,	10.9 kg/thm reduction in CR.	(32)
350 m ³ /thm of H ₂	CR reduced from 300 kg/thm to 223.6 kg/thm.	(30)
300 m ³ /thm of H ₂	CR reduced from 358 (with 167 PCI) to 285 (with 167 PCI).	(20)
10 m ³ /thm of H ₂	CR reduced by 0.48%/0.46% at a constant/variable RAFT.	(39)
Increase H ₂ injection from 10 kg/thm to 35 kg/thm	CR decreased by 60 kg/thm	(38)
HI (injection of 10 kg/thm) at 80 °C	CR 498.1 to 461.7 kg/thm	(48)
Increase in coke oven gas (COG) injection rate from 14.8 kg/thm (current) to 74.0 kg/thm, NG from 0.0 kg/thm (current) to 43.0 kg/thm and increase in H ₂ injection rate from 0.0 kg/tHM (current) to 12.0 kg/tHM.	Reduction in CR from 308.1 kg/thm (current) to 277.1 kg/thm (COG), 272.1 kg/thm (NG) and 286.3 kg/thm (H ₂).	(51)
Injection of 30 m ³ /s of hydrogen (shaft injection).	Reduction in CR from 323 kg/thm to 295 kg/thm	(56)
Co-injection of H ₂ reduced the coke rate.	The lowest CR (285.8 kg-C/thm) and the highest coke replacement rate (4.1 kg-C/kg-H ₂) being achieved with an H ₂ injection ratio of 4:6 in the tuyere and shaft.	(5)
With a shaft injection rate of 10 Nm ³ /s.	The fuel rate decreased from 503 kg/thm to 445 kg/thm.	(57)

Source: Prepared by the authors themselves

IV. Combustion Efficiency and Substitution Rate

Hydrogen injection slightly improves pulverized coal burn (PCI) up to 30 m³/thm, but above that the burn rate decreases (15). Hydrogen combustion is more efficient and faster than PCI, being completely consumed 0.5 m from the tip of the injection lance (34). The increase in H₂ in the co-injected gas speeds up its combustion, heats up the PCI and anticipates its devolatilization and combustion, compensating for the capture of O₂ by the gas (58). The combustion of H₂ raises the local temperature, promoting the early release of volatiles and greater burnout of the biochar (59). Increasing the injection of natural gas (NG) increases the fraction of H₂, reducing CO₂ emissions, but also O₂ in the raceway, impairing the burning of PCI. The PCI burn rate fell from 70.11% to 68.31% when NG injection was increased from 47 to 67 m³/thm (60).

The combustion efficiency of PCI is affected by the consumption of O₂ by fuels rich in H₂. Burnout was highest with recycling top gas (RTG) (81.76%), followed by H₂ (79.86%), COG (78.04%) and NG (75.72%). NG generated more unburned particles, affecting the permeability of the coke bed (61). Burnout increases along the centerline as the distance from the injection lance extends and is greater around the particle plume. With high NG injection and low PCI, the burnout at the exit of the combustion zone ranges from 64.12% to 68.11% (62). Both the final burnout along the centerline of the vent and the average burnout of the raceway surface increased with increasing hydrogen injection rate (63).

The PCI substitution rate varies between 1.9 kg-PCI/kg-H₂ and 1.64 kg-coke/kg-H₂ for HI of 19 kg/thm (44). Coke replacement ranges from 4 kg/kg-H₂ (with 5 kg/thm of H₂) to 3 kg/kg-H₂ (maximum injection) (35). With 69 kg/thm of H₂, the total consumption of reducing agent was 376 kg/thm (307 kg of coke), compared to 332 kg of coke and 150 kg of PCI in conventional BF (64). The substitution ratio ranges from 0.86 to 5.49, with the highest value for 5 kg/thm of H₂ at 1200 °C (48). Typical substitutions: NG (0.8-0.9 kg/m³), COG (0.4-0.45 kg/m³), H₂ (1.9-2.0 kg/kg) (65).

Blowpipe injection is ideal for small quantities and low O₂ contents (up to 30%), improving PCI burnout. Direct injection into the raceway is more suitable for large volumes and high O₂ contents (40-50%), promoting diffusion, temperature increase and burnout (66). Smaller PCI particles perform better. However, large volumes of reducing gas can cause tuyere erosion, requiring further study.

V. Impact on primary slag and the Softening and Melting Zone

The injection of hydrogen-rich gas reduces the FeO content and the amount of primary slag. With 10% H₂ in the reducing gas, the slag decreased by around 10% (01). The FeO content fell to 1.73%, below the 10-30% of conventional operation, and the cohesive zone took on an irregular V shape, descending to the bosh (67). Enrichment with H₂ alters slag formation, reducing the amount of wustite and forming new mineral phases. Without H₂, the slag contained wustite, hortonolite and FeTiO₃; with 10% H₂, the wustite disappeared and calcium-magnesium olivine and melilite appeared. The amount of primary slag fell with increasing H₂: 1290 kg/tFe (0%), 919 kg/tFe (5%), 618 kg/tFe (10%) and 512 kg/tFe (15%) (37).

With more H₂, the reduction rate of the ore increases, reducing the FeO content and raising the slag formation temperature, which increases the initial softening temperature (T_A). The continuous reduction of FeO to Fe raises the softening end temperature (T₄₀) and the melting start temperature (T_S) of the mixed charge. The melting point of the slag also increases with the H₂ content (68). In the oxygen blast furnace (OBF) process with H₂, the reduction index (RI) must be improved, and the T_A can be reduced to 1050-1100 °C, while in conventional BF, the T_A can exceed 1100-1150 °C (64).

The injection of H₂ can improve the permeability of the cohesive zone due to the lower presence of wustite and the higher melting point of the slag (23). With 15% H₂, T_S increased from 1305 °C to 1485 °C and the dripping temperature (T_D) from 1413 °C to 1497 °C (69). The cohesive zone becomes narrower horizontally and shorter vertically with more H₂ (40). The formation of high melting point compounds and the carburization of iron raised the dripping temperature, shifting the cohesive zone downwards (37). The height of the cohesive zone fell below the baseline with 23.75 kg/thm of H₂ (38).

With 15% H₂, T_A rose from 1112 °C to 1176 °C, T₄₀ from 1220 °C to 1377 °C, T_S from 1323 °C to 1465 °C, and the width of the cohesive zone fell from 141 °C to 27 °C (68). With the injection of H₂ into the shaft, the apex of the cohesive zone rose, the base remained stable and the profile became more concave towards the shaft (52), this can be considered positive for the process because it improves thermal and reduction conditions, accelerates reduction, and increases the degree of reduction of iron oxides. These changes reduce fuel consumption.

VI. Productivity

Productivity increased by 145% (from 4000 to ~9800 thm/day) with the option of maintaining the RAFT by enriching the blowing air with oxygen, when the HI went from 0 to

585 m³/thm. The increase in productivity was less significant when lower RAFTs were allowed. Compared to other gases, HI resulted in a 48% increase in productivity, natural gas in 30% and coke oven gas in 14%, while maintaining the RAFT between 1800 and 2200 (39). Hydrogen injection increased blast furnace productivity by up to 13.7%, depending on the level of oxygen enrichment (40). BF productivity increases with hydrogen enrichment, but the rate of increase decreases at higher hydrogen contents (37). Productivity increases with increasing oxygen content in the blow, but the increase is less significant in oxygen blast furnaces (OBFs) with injection in the tuyeres due to the reduction in the blow rate. (53). Partially replacing PCI with enriched blast furnace gas (with H₂) resulted in a significant increase in productivity, with the best-case scenario predicting an increase of 20.5% (70).

VII. Combustion Zone

The combustion zone (CZ), or raceway, can be altered in size with hydrogen injection (HI), but there is disagreement in the literature as to its actual extent.

Some studies indicate that the CZ is smaller with HI compared to pulverized coal injection (PCI), due to the lower density of hydrogen. This characteristic reduces the drag force on the coke particles, limiting the expansion of the CZ and favoring a more peripheral distribution of the gases generated. It was observed that the width and height of the CZ gradually decreased by around 32% and 55%, respectively, as the concentration of injected hydrogen increased from 1.5% to 4.5%. The hydrogen burns near the injection nozzle, releasing a large amount of water vapor, and the oxygen is quickly consumed due to the simultaneous combustion of hydrogen and carbon. As a result, CO is formed by the gasification of carbon with H₂O and CO₂ on the periphery of the CZ (71). On the other hand, other studies point to the opposite behavior, suggesting that HI results in a significantly larger CZ than that observed with PCI. According to these studies, the depth of the CZ expands from 1.1 m to 1.4 m, the width from 0.45 m to 0.7 m and the height from 0.76 m to 1.45 m. The total volume of the SZ under HI can reach 0.19 m³, around four times greater than that of the PCI (0.05 m³). In this condition, a higher temperature is observed at the front of injection lance, but lower temperatures in deeper regions of the CZ. This effect is attributed to the intense endothermic reaction of coke gasification, promoted by the large amount of water vapor generated during hydrogen combustion (34).

The gas velocity in the tuyere varies significantly with the type of fuel injected. The decreasing order of speed is: COG (313.89 m/s), NG (308.13 m/s), RTG (297.08 m/s) and H₂ (282.14 m/s), and the speed of the gas affects penetration into the coke bed. COG, due to its

higher velocity, has the highest penetration, while H₂ has the lowest (61). Injecting H₂ can reduce carbon emissions, but requires additional measures to increase the gas velocity in the combustion zone.

VIII. Reducing CO₂ emissions

One of the main drivers of hydrogen injection in BF_s is CO₂ mitigation. Studies report reductions of up to 20–30%, depending on injection rate and operating conditions. The benefits, however, depend on H₂ origin: if produced from fossil fuels without CCS, overall emissions may not fall significantly (1). Thus, integration with renewable sources or CCS is essential. Table 03 summarizes published results.

Additional findings include: reformed COG injection reduced CO₂ by 34.62%; higher H₂ content raises BFG calorific value, improving energy recovery and lowering emissions (56); combining H₂ injection with TGT recycling cuts CO₂ by up to 30%, and potentially 100% with CCS (75); COG and pure H₂ achieved ~8% and 6% reductions, respectively, versus conventional BF (51); replacing PCI with enriched BFG reduced emissions by up to 14.8% (70); and H₂ injection achieved 13.03 kgCO₂/kgH₂ avoided (76).

Table 3 - Potential Reduction in CO₂ Emissions

Potential Reduction	H ₂ injection	Reference
16% reduction	359 Nm ³ /thm	(16)
Reduction of CO ₂ emissions by up to 262 m ³ /thm (an OE of the blowing air of 9% and integrating CO ₂ capture, the final emission was 278 m ³ /thm of CO ₂).	80 m ³ /thm	(43)
Direct CO ₂ reduction of 10.2%-17.8% (decarbonization potential was estimated in the range of 9.4 to 9.7 tCO ₂ /tH ₂).	HI 19.0 and 28.3 kg/thm.	(35)
HI reduced direct emissions by 8% with a total emissions reduction of 110 kg-CO ₂ eq/thm.		(44)
14.4% reduction in CO ₂ emissions.	359 m ³ /thm	(45)
26.1% reduction in CO ₂ emissions .	350 m ³ /thm	(32)
Can reduce CO ₂ emissions by 16%, 7% and 6%	300, 200 e 100 m ³ /thm (H ₂ , COG and NG).	(20)
9.46% reduction in CO ₂ emissions.	19.5 kg/thm	(54)
21.3%/25.9% reduction in emissions at a constant/variable RAFT [Every 10 m ³ of H ₂ reduces CO ₂ production by 0.49 and 0.51% (constant and variable RAFT)].	The increase in H ₂ consumption from 150/100 to 585/600 m ³ /thm	(39)
Potential 11% reduction in CO ₂ emissions.	Increase in H ₂ from 10 to 35 kg/thm	(38)
Potential reduction of between 5.5-12.1%.	27.5 kg/thm of H ₂	(72)
21.4% reduction in CO ₂ emissions compared to the reference case using 120 kg/thm of PC.	27.5 kg/thm of H ₂ are injected at 1200 °C	(48)
Reduction of CO ₂ emissions by 82 kg/thm, representing 5.4%.	Co-injection of 95 kg of NG with 15 kg of H ₂	(73)
Reduced emissions by 9.8% (from 1397 kg/thm to 1259 kg/thm)	29 kg/thm of H ₂	(74)

Source: Prepared by the authors themselves

IX. Other operational impacts

The injection of hydrogen reduced the Maximum Pressure Difference (MPD) in all the volume fractions tested, indicating an improvement in the permeability of the process (69), and at higher rates with 35 kg/thm of H₂, the pressure drop across the furnace increased significantly (38). The S value (permeability index) decreased significantly with increasing hydrogen content, from 238.78 to 19.29 kPa °C (68). The increase in hydrogen content in the reducing gas reduced the pressure drop in the furnace, despite the decrease in the coke layer (33).

Stable blast furnace operation can be achieved with H₂ injection of up to 30 kg/thm alone and up to 15 kg/thm in combination with NG (68). For a blast temperature of 1200°C, the maximum hydrogen injection rate, limited by the value of $\Delta t_{min} = 20^{\circ}\text{C}$, is 19.0 kg of H₂/thm for 2% oxygen enrichment and 28.3 kg of H₂/thm for 12% enrichment (35). The H₂ residence time decreases with increasing HI and with increasing injection temperature (77).

X. Raw materials

The addition of hydrogen to the reducing gas significantly increases the reduction rate of iron ore above 600-700 °C for all charges (lump, sinter and pellets) (78). Mixing H₂ with CO improves gas diffusion, promoting deeper reduction in the particles and the formation of cracks, which can intensify degradation. Just 2% of H₂ in CO significantly increases RDI-2.8, especially for sinter (50% increase) and pellets (137% increase) (79).

In blast furnaces, at around 900 °C, H₂ favors the reduction of pellets, with the formation of radial cracks that facilitate the penetration of the reducing gas (80). The presence of H₂/H₂O improves reduction kinetics without causing significant disintegration (26). In the sinter, H₂ can inhibit or intensify disintegration, depending on the kinetics in relation to the gas mixture. Reduction with H₂ is kinetically more favorable than with CO, forming dense, interconnected layers of iron, while CO generates smaller fragments (23).

Coke is 20-30% more reactive with H₂O than with CO₂, and the reaction rate is 2-4 times higher (81). The H₂ generated by the injection of hydrogen-rich gas promotes the coke gasification reaction, increasing the solution loss reaction rate (82). When the fraction of H₂ in the oven increases from 0 to 10%, the coke gasification rate increases significantly. Gasification with H₂O is 1.3-3.2 times faster than with CO₂, with a 37 °C lower limit temperature (54). The addition of 10% and 25% H₂ increases coke gasification from 27.05% to 33.98% and 42.83%, respectively, and reactivity from 22.3% to 28.4% and 35.8% (83). The effect of H₂ on gasification is more relevant at low temperatures, becoming negligible above 1300 °C (84). H₂O reacts

preferentially with coke up to 1092 K, shifting gasification to lower temperatures (85).

H₂-rich blast furnaces alter the coke's deterioration mechanism, promoting the reduction process in low-temperature areas and lower production of H₂O and CO₂ at high temperatures. The strength of the coke after reaction drops from 66.4% to 59.4% with 10% H₂, and to 50.4% with 25% (83). The addition of H₂O increases the formation of micropores at low temperatures and macropores at high temperatures, with greater porosity at the edges of the coke compared to CO₂ (81). With 10% H₂, the porosity of the coke rises from 57.9% to 83%, while that of the center drops to 40% (83).

XI. Future research directions

Future research should prioritize the optimization of hydrogen injection into blast furnaces (BFs), with a focus on developing advanced models capable of predicting its operational effects, as well as investigating new materials and technologies for storage and injection. In addition, it is essential to carry out industrial-scale studies to assess the long-term performance of this technology, especially with regard to furnace stability and the quality of the end product. Several technical challenges still need to be overcome, such as recycling the top gas containing N₂, the operational difficulties associated with high levels of oxygen in the blowing air and the limitation of the indirect reduction rate in conventional blast furnaces, currently restricted to 70%, which compromises the potential for reducing carbon emissions (86).

Direct hydrogen injection has limitations, including low utilization efficiency, difficulties in the thermal balance and safety concerns regarding handling and storage. In this context, the methanation of BFG, a process that converts CO and CO₂ into CH₄, has emerged as a promising alternative for the indirect introduction of hydrogen, with superior potential for mitigating CO₂ emissions (74).

The integration of hydrogen injection with low-carbon technologies, such as carbon capture and storage (CCS) and the use of renewable energy sources, represents a relevant complementary strategy. In an integrated steel plant with capacity of 2.54 million tons/year, the estimated cost of capturing 99% of the CO₂ is \$80.3 per ton (87).

Other aspects that require investigation include optimizing the H₂/CO₂ ratio in the methanation process ($\text{CO}_2 + 4\text{H}_2 = \text{CH}_4 + 2\text{H}_2\text{O}$ and $\text{CO} + 3\text{H}_2 = \text{CH}_4 + \text{H}_2\text{O}$) to maximize coke replacement (88), and developing materials that are resistant to hydrogen embrittlement. In addition, the H₂/CO ratio must be adjusted to optimize NO reduction and minimize NO_x emissions, since excess hydrogen can compromise the regeneration of the active sites in the coke,

reducing the efficiency of the process (89).

Finally, the success of the transition to hydrogen-based steelmaking will depend on the availability of green hydrogen, the expansion of renewable electricity and the necessary infrastructure. It is estimated that, in the most ambitious scenarios, global electrolysis capacity will need to be increased by more than a thousand times compared to current levels (90).

CONCLUSION

The injection of hydrogen into blast furnaces represents a promising way to reduce CO₂ emissions in the steel industry while maintaining production efficiency. However, successful implementation requires careful optimization of operating parameters, as well as the development of new materials and technologies to overcome technical challenges. This technology in blast furnaces shows potential for reducing CO₂ emissions by more than 30%, as well as reducing coke consumption by more than 10% and increasing productivity, but it requires strict control of parameters such as RAFT, oxygen enrichment, top gas temperature, combustion and cohesive zone shape and raw material quality. Given the current production costs of the entire green hydrogen chain, replacing coal and coke is not financially advantageous. Efforts from the government, academia, and industry are needed to find ways to overcome these difficulties.

This review has highlighted the main impacts and applications of hydrogen injection in BFs, including its effects on temperature profiles, gas composition, coke consumption and CO₂ emissions. While there has been significant progress in understanding the potential of hydrogen injection, more research is needed to address the remaining challenges and unlock the full potential of this technology. By continuing to advance and understand the production of iron from hydrogen, we can move closer to achieving a sustainable, low-carbon future for the steel industry.

REFERENCES

1. Fan Z, Friedmann SJ. Low-carbon production of iron and steel: Technology options, economic assessment, and policy. Vol. 5, Joule. Cell Press; 2021. p. 829-62.
2. Shatokha V. A Study on the Potential of Carbon Dioxide Utilization Through its Co-Injection with Hydrogen into the Blast Furnace Tuyeres. Steel Res Int. 1^o, November 2023;
3. Energy Agency I. Iron and Steel Technology Roadmap Towards more sustainable steelmaking Part of the Energy Technology Perspectives series [Internet]. Available at: www.iea.org/t&c/

4. Zeji Tang S, Zheng Z, Chen H, He K, The Influence of Hydrogen Injection on the Reduction Process in the Lower Part of the Blast Furnace: A Thermodynamic.
5. Zhao Z, Yu X, Li Y, Zhu J, Shen Y. CFD study of hydrogen co-injection through tuyere and shaft of an ironmaking blast furnace. *Fuel*. September 15, 2023;348.
6. Hasanbeigi A. Steel Climate Impact An International Benchmarking of Energy and CO₂ Intensities [Internet]. 2022. Available at: <https://www.globalefficiencyintel.com>
7. Mathieson JG, Rogers H, Somerville MA, Jahanshahi S. Reducing net CO₂ emissions using charcoal as a blast furnace tuyere injectant. *ISIJ International*. 2012;52(8):1489-96.
8. Leão AS, Medeiros DL, Santiago MA, do Carmo Tavares AO, Maranduba HL, dos Santos Almeida E. Rigorous environmental and energy life cycle assessment of blast furnace pig iron in Brazil: The role of carbon and iron sources, and co-product utilization. *Sustainable Materials and Technologies*. July 1^o, 2023;36.
9. de Castro JA, de Medeiros GA, de Oliveira EM, Nogami H. A Comprehensive Modeling as a Tool for Developing New Mini Blast Furnace Technologies Based on Biomass and Hydrogen Operation. *Journal of Sustainable Metallurgy*. June 1^o, 2020;6(2):281-93.
10. Bailera M, Lisbona P, Peña B, Romeo LM. A review on CO₂ mitigation in the Iron and Steel industry through Power to X processes. Vol. 46, *Journal of CO₂ Utilization*. Elsevier Ltd; 2021.
11. Castro JA De, Takano C, Yagi JI. A theoretical study using the multiphase numerical simulation technique for effective use of H₂ as blast furnaces fuel. *Journal of Materials Research and Technology*. July 1^o, 2017;6(3):258-70.
12. Yokoyama H, Kasai E, Naitou Masaaki. Reduction of CO₂Emissions from Blast Furnace Applying Reactive Coke Agglomerate and Hydrogen Reduction. *Tetsu- to-Hagané Journal of the Iron and Steel Institute of Japan*. 2022; Vol. 108, No. 6, pp. 335-342.
13. Chen Y, Zuo H. Review of hydrogen-rich ironmaking technology in blast furnace. Vol. 48, *Ironmaking and Steelmaking*. Taylor and Francis Ltd; 2021. p. 749-68.
14. Tang J, Chu M, Li F, Zhang Z, Tang Y, Liu Z, et al. Mathematical simulation and life cycle assessment of blast furnace operation with hydrogen injection under constant pulverized coal injection. *J Clean Prod*. 1^o January 2021;278.
15. Wang Z, Zhang J, Xie C, Gao L, He Z, Zhan W. Numerical Simulation Study of Co-injection of Pulverized Coal and Hydrogen in the Tuyere-Raceway. *Journal of Sustainable Metallurgy*. September 1^o, 2024;10(3):1237-49.
16. Kamijo C, Matsukura Y, YOKOYAMA H, SUNAHARA K, KAKIUCHI K, SAKAI H, et al. Influence of Large Amount of Hydrogen Containing Gaseous Reductant Injection on Carbon Consumption and Operation Conditions of Blast Furnace - Development of Low Carbon Blast Furnace Operation Technology by using Experimental Blast Furnace: part II -. *ISIJ International*. December 1^o, 2022;62(12):2433-41.

17. Li H, Zhao Y, Guo C, Li J. Analysis of Technological Pathways and Development Suggestions for Blast Furnace Low-Carbon Ironmaking. *Metals* (Basel). 2024 Nov 1;14(11).
18. Chang Y, Wan F, Yao X, Wang J, Han Y, Li H. Influence of hydrogen production on the CO₂ emissions reduction of hydrogen metallurgy transformation in iron and steel industry. *Energy Reports*. December 1^o, 2023;9:3057-71.
19. Martin P, Ocko IB, Esquivel-Elizondo S, Kupers R, Cebon D, Baxter T, et al. A review of challenges with using the natural gas system for hydrogen. *Energy Sci Eng*. 2024 Oct 1;
20. Chaika AL, Lebed' V V., Kornilov B V., Moskalina AA, Karikov SA. Heat and Power Analysis of Technologies for Reducing Carbon Dioxide Emissions and Increasing the Energy Efficiency of Blast-Furnace Production. *Steel in Translation*. January 1, 2021;51(1):68-72.
21. Pawelec G, Fonseca J. Steel from Solar Energy: A Techno-Economic Assessment of Green Steel Manufacturing. Brussels: Hydrogen Europe; 2022 [cited 2025 Jun 10]. Available from: <https://www.hydrogeneurope.eu>
22. Argus Media Group. Steel: What is the cost of going green? Insight Paper. February 2025 [Internet]. Available from: <https://www.argusmedia.com/pt/news- and-insights/market-insight-papers/steel-what-is-the-cost-of-going-green>
23. Lan C, Hao Y, Shao J, Zhang S, Liu R, Lyu Q. Effect of H₂ on Blast Furnace Ironmaking: A Review. Vol. 12, *Metals*. MDPI; 2022.
24. Buchwalder J, Mernitz J, Harp G, Reuther C, Schingnitz M. Development of injection of reduction gas into the blast furnace shaft.
25. Higuchi K, Matsuzaki S, Saito K, Nomura S. Improvement in reduction behavior of sintered ores in a blast furnace through injection of reformed coke oven gas. *ISIJ International*. October 15, 2020;60(10):2218-27.
26. Ahmed H, Sideris D, Björkman B. Injection of H₂-rich carbonaceous materials into the blast furnace: devolatilization, gasification and combustion characteristics and effect of increased H₂-H₂O on iron ore pellets reducibility. *Journal of Materials Research and Technology*. November 1^o, 2020;9(6):16029- 37.
27. Maier C, Jordan C, Harasek ME, Feilmayr C, Thaler C. Investigation of alternative reducing agent injection into the raceway of blast furnaces using CFD. In: *Chemical Engineering Transactions*. Italian Association of Chemical Engineering - AIDIC; 2013. p. 1345-50.
28. Qin L, Fang J, Zhu S, Zhao B, Han J. Municipal solid waste gasification by hot recycling blast furnace gas coupled with in-situ decarburization to prepare blast furnace injection of hydrogen-rich gas. *Waste Management*. February 15, 2024;174:153-63.
29. Tang J, Chu M sheng, Li F, Feng C, Liu Z gen, Zhou Y sheng. Development and progress on hydrogen metallurgy. Vol. 27, *International Journal of Minerals, Metallurgy and Materials*. University of Science and Technology Beijing; 2020. p. 713-23.

30. Fucheng Z, Lukuo H, Ying X. Prospects for green steelmaking technology with low carbon emissions in China. Vol. 6, Carbon Energy. John Wiley and Sons Inc; 2024.
31. Zhang Y, Liu J, Wang Y, Li X, Zhao Y, Chen H. Industrialized application of hydrogen-rich gas injection into blast furnace of Jinnan Steel. *Ironmaking & Steelmaking*. 2023;50(9):1123-31.
32. Ye S, Xie Z, Wang Y, Wu W, Wang R, Zhang Y, et al. New insights into the influence of hydrogen on important parameters of blast furnace. *J Clean Prod*. 1^o November 2023;425.
33. Nogami H, Kashiwaya Y, Yamada D. Simulation of blast furnace operation with intensive hydrogen injection. *ISIJ International*. 2012;52(8):1523-7.
34. Zhuo Y, Hu Z, Shen Y. CFD study of hydrogen injection through tuyeres into ironmaking blast furnaces. *Fuel*. October 15, 2021;302.
35. Shatokha V. Modeling of the effect of hydrogen injection on blast furnace operation and carbon dioxide emissions. *International Journal of Minerals, Metallurgy and Materials*. October 1^o, 2022;29(10):1851-61.
36. Barrett N, Mitra S, Doostmohammadi H, O'Dea D, Zulli P, Chew S, et al. Development of softening and melting testing conditions simulating blast furnace operation with hydrogen injection. *Ironmaking and Steelmaking*. 2023;50(9):1248-59.
37. Qie Y, Shangguan D, Li Y, Wang X, Lyu Q, Wang X. Formation of Primary Slag and Carburizing Behavior of Metal Iron in Cohesive Zone of Hydrogen-rich Blast Furnace. *ISIJ International*. 2024;64(9):1360-6.
38. Okosun T, Nielson S, Zhou C. Blast Furnace Hydrogen Injection: Investigating Impacts and Feasibility with Computational Fluid Dynamics. *JOM*. 2022;
39. Chaika OL, Kornilov B V., Myravyova IG, Moskalyna AO, Lebid V V., Ivancha NG. PROSPECTS FOR THE USE OF HYDROGEN AND HYDROGEN- CONTAINING ADDITIVES TO REDUCE CO₂ EMISSIONS AND TO IMPROVE THE PERFORMANCE OF BLAST FURNACE SMELTING. *Science and Innovation*. 2024;20(5):35-52.
40. Li J, Kuang S, Jiao L, Liu L, Zou R, Yu A. Numerical modeling and analysis of hydrogen blast furnace ironmaking process. *Fuel*. September 1^o, 2022;323.
41. Immonen J, Powell KM. Dynamic modeling of a direct reduced iron shaft furnace to enable pathways towards decarbonized steel production. *Chem Eng Sci*. 2024 Dec 5;300.
42. Chen B, Chen J, Bai C, Hu M, Chen M. Optimization of Raceway Adiabatic Flame Temperature Model for H₂-Rich Gas Injection Blast Furnace. *Journal of Sustainable Metallurgy*. December 1^o, 2024;
43. Shatokha V. Evaluation of Some Uncertainties in the Modeling of Blast Furnace Hydrogen Injection. *Steel Res Int*. 2024;
44. Wang X, Qie Y, Lü Q, Wang Y. Analysis and comparison of energy saving and emission

reduction of blast furnace injection with different hydrogen-rich mediums. *Iron & Steel*. 2024;60(4):45-56.

45. Gao X, Zhang R, You Z, Yu W, Dang J, Bai C. Use of Hydrogen-Rich Gas in Blast Furnace Ironmaking of V-bearing Titanomagnetite: Mass and Energy Balance Calculations. *Materials*. September 1^o, 2022;15(17).

46. Hu Y, Qiu Y, Chen J, Hao L, Rufford TE, Rudolph V, et al. Integrating a Top-Gas Recycling and CO₂ Electrolysis Process for H₂-Rich Gas Injection and Reduce CO₂ Emissions from an Ironmaking Blast Furnace. *Materials*. March 1^o, 2022;15(6).

47. Barrett N, Mitra S, Doostmohammadi H, O'Dea D, Zulli P, Chew S, et al. Assessment of Blast Furnace Operational Constraints in the Presence of Hydrogen Injection. *ISIJ International*. 2022;62(6):1168-77.

48. Yilmaz C, Wendelstorf J, Turek T. Modeling and simulation of hydrogen injection into a blast furnace to reduce carbon dioxide emissions. *J Clean Prod*. June 15, 2017;154:488-501.

49. Jiao K, Feng G, Zhang J, Wang C, Zhang L. Effect of multi-component gases on the behavior and mechanism of carbon deposition in hydrogen-rich blast furnaces. *Energy*. January 15, 2023;263.

50. Okosun T, Nielson S, D'Alessio J, Ray S, Street S, Zhou C. On the impacts of pre-heated natural gas injection in blast furnaces. *Processes*. July 1^o, 2020;8(7).

51. Mauret F, Baniasadi M, Saxén H, Feiterna A, Hojda S. Impact of Hydrogenous Gas Injection on the Blast Furnace Process: A Numerical Investigation. *Metallurgical and Materials Transactions B: Process Metallurgy and Materials Processing Science*. August 1^o, 2023;54(4):2137-58.

52. Rahmatmand B, Tahmasebi A, Lomas H, Honeyands T, Koshy P, Hockings K, et al. A technical review on coke rate and quality in low-carbon blast furnace ironmaking. Vol. 336, *Fuel*. Elsevier Ltd; 2023.

53. Li J, Kuang S, Zou R, Yu A. Numerical Investigation of Burden Distribution in Hydrogen Blast Furnace. *Metallurgical and Materials Transactions B: Process Metallurgy and Materials Processing Science*. December 1^o, 2022;53(6):4124- 37.

54. Chang C, Song H, Lei F, Di Z, Cheng F. Study on the carbon reduction potential of blast furnace injection process using reformed coke oven gas. *Huagong Xuebao/CIESC Journal*. June 25, 2024;75(6):2344-52.

55. Bahgat M, Abdel Halim KS, El-Kelesh HA, Nasr MI. Blast furnace operating conditions manipulation for reducing coke consumption and CO₂ emission. *Steel Res Int*. July 2012;83(7):686-94.

56. Yu X, Hu Z, Shen Y. Modeling of hydrogen shaft injection in ironmaking blast furnaces. *Fuel*. October 15, 2021;302.

57. Jiao L, Nie H, Kuang S, Li J, Yu A. Numerical simulation and analysis of oxygen blast

furnace under different injection conditions. *Fuel*. August 1^o, 2024;369.

58. Ren M, Liu W, Zhao J, Zou C, Ren L, Wu H, et al. Effects of hydrogen fraction in co-injection gas on combustion characteristics of the raceway in low carbon emission blast furnace. *Int J Hydrogen Energy*. April 8, 2023;48(30):11530-40.

59. Gan MJ, Liu Y, Shen Y. A novel ironmaking decarbonization technology - co- injection of hydrogen and biochar (CoHB): A CFD study of combustion in the raceway under simulated blast furnace conditions. *Fuel*. October 15, 2023;350.

60. Zhang C, Vladislav L, Xu R, Sergey G, Jiao K, Zhang J, et al. Blast furnace hydrogen-rich metallurgy-research on efficiency injection of natural gas and pulverized coal. *Fuel*. March 1^o, 2022;311.

61. Zhang C, Zhang J, Zheng A, Xu R, Jia G, Zhu J. Effects of hydrogen-rich fuel injection on the states of the raceway in blast furnace. *Energy*. July 1^o, 2023;274.

62. Zhang C, Zhang J, Xu R, Zheng A, Zhu J, Li T. Numerical investigation of hydrogen-rich gas and pulverized coal injection in the raceway of a blast furnace with lower carbon emissions. *Fuel*. January 15, 2024;356.

63. Liu Y, Hu Z, Shen Y. CFD Study of Hydrogen Injection in Blast Furnaces: Tuyere Co-injection of Hydrogen and Coal. *Metallurgical and Materials Transactions B: Process Metallurgy and Materials Processing Science*. October 1^o, 2021;52(5):2971-91.

64. Wang P, Wu Y, Meng Q min, Di Z xia. Interaction reaction between the degradation of coke and the reduction of iron ore in oxygen blast furnace under hydrogen-enriched operation. *Ironmaking and Steelmaking*. March 15, 2020;47(3):290-5.

65. Babich A. Blast furnace injection for minimizing the coke rate and CO₂ emissions. Vol. 48, *Ironmaking and Steelmaking*. Taylor and Francis Ltd; 2021. p. 728-41.

66. Peng X, Wang J, Li C, Zuo H, Wang G, She X, et al. Influence of Reducing Gas Injection Methods on Pulverized Coal Combustion in a Medium Oxygen- Enriched Blast Furnace. *JOM*. October 1^o, 2021;73(10):2929-37.

67. Wang F, Ye S, Yang P, Qi M, Zhang Y, Wu W, et al. Continuous reduction and phase transformation mechanism of pellets in lumpy zone based on dissected hydrogen blast furnace. *Int J Hydrogen Energy*. September 26, 2024;84:580-92.

68. Chen Z, Ye S, Geng S, Zhu K, Zhang Y, Zou X, et al. Softening and melting performance of mixed burden under the simulated hydrogen-rich blast furnace atmosphere. *Ironmaking and Steelmaking*. 2023;50(5):538-47.

69. Xiao Y, Zhu K, Ye S, Xie Z, Zhang Y, Lu X. Hydrogen on softening-melting and slag forming behavior under the operation of blast furnace with iron coke charging. *Int J Hydrogen Energy*. August 26, 2022;47(73):31129-39.

70. de Castro JA, de Medeiros GA, da Silva LM, Ferreira IL, de Campos MF, de Oliveira EM. A Numerical Study of Scenarios for the Substitution of Pulverized Coal Injection by Blast

Furnace Gas Enriched by Hydrogen and Oxygen Aiming at a Reduction in CO₂ Emissions in the Blast Furnace Process. *Metals* (Basel). May 1^o, 2023;13(5).

71. E D, Zhou P, Ji L, Cui J, Xu Q, Guo L, et al. Particle-scale modelling of injected hydrogen and coke co-combustion in the raceway of an ironmaking blast furnace. *Fuel*. 2023 Mar 15;336.

72. Ma K, Deng J, Wang G, Zhou Q, Xu J. Utilization and impacts of hydrogen in the ironmaking processes: A review from lab-scale basics to industrial practices. Vol. 46, *International Journal of Hydrogen Energy*. Elsevier Ltd; 2021. p. 26646-64.

73. Ugarte OJ, Okosun T, Nielson S, Zhou CQ. Correlating Impacts of Injected Fuels on Carbon Emissions in Blast Furnace with Computational Fluid Dynamics Modeling. *Steel Res Int*. December 1^o, 2023;94(12).

74. Aubry N, Estabrooks M, Liu F, Perez D, Elliott R, Sukhram M, Baumann P, Cameron I. Methanation - An opportunity to recycle carbon and more efficiently introduce hydrogen to the blast furnace. *Iron & Steel Technology*. April 2025 [Internet];2025(Apr):50-60. Available from: https://www.aist.org/getmedia/1dd100d9-bfac-4277-89f2-110c9c16abd2/April-2025_050-060.pdf

75. Shahabuddin M, Brooks G, Rhamdhani MA. Decarbonization and hydrogen integration of steel industries: Recent development, challenges and technoeconomic analysis. Vol. 395, *Journal of Cleaner Production*. Elsevier Ltd; 2023.

76. Wu W, Liang JL, Chen YM, Shiau JS, Lin BJ, Kuo PC. Development of simple blast furnace models for addressing carbon reduction strategies. *Energy Convers Manag*. February 15, 2024;302.

77. Yu X, Shen Y. CFD study of the residence time distribution of shaft-injected hydrogen in a blast furnace. *Fuel*. September 15, 2024;372.

78. Barrett N, Mitra S, Chew S, O'Dea D, Honeyands T. Effect of Hydrogen Addition on Softening and Melting Performance of Lump and Sinter Mixed Burden. *ISIJ International*. 2023;63(10):1626-36.

79. Barustan MIA, Copland E, Nguyen TBT, O'Dea D, Honeyands T. Reduction Degradation of Lump, Sinter, and Pellets in Blast Furnace with Hydrogen Injection. *ISIJ International*. August 1^o, 2024;64(10):1517-27.

80. Xu C, Xu A, Zhang J, Liu Z, Wang Y, Wang R, et al. Effect of CO-CO₂ and H₂- H₂O on the reduction degree of fluxed pellets: reduction mechanism within hydrogen-rich blast furnace. *Ironmaking and Steelmaking*. 2022;49(9):932-9.

81. Zhou F, Peng D, Li K, Conejo AN, Liao H, Xiong Z, et al. Coke behavior with H₂O in a hydrogen-enriched blast furnace: A review. Vol. 31, *International Journal of Minerals, Metallurgy and Materials*. University of Science and Technology Beijing; 2024. p. 959-76.

82. Zhang S., Wu X., Liu R., Lan C., Yao H., Ji H. Discussion on smelting characteristics

and problems of blast furnace with hydrogen-rich gas injection. Vol. 59, Kang T'ieh/Iron and Steel. January 2024; p. 1-11.

83. Fan X, Li C, Wang M, Liu Y, Feng Y, Liu Y, et al. Effects of Adding Different Proportions of H₂ to the Simulated Hydrogen-Rich Blast Furnace. Chem Eng Technol. December 1^o, 2022;45(12):2284-91.

84. Golubev O V., Chernousov PI, Myasoedov S V., Karataeva A V. Review of Recent Research on Coke Gasification in a Blast Furnace with Elevated Hydrogen Levels and Determination of the Coke's Postreactive Strength CSR 2. Laboratory Research on Coke Gasification in the Presence of H₂ and H₂O. Coke and Chemistry. February 1^o, 2024;67(2):86-93.

85. Golybev O V., Chernousov PI, Demidov AY, Slonov M V. Review of Recent Research on Coke Gasification in a Blast Furnace with Elevated Hydrogen Levels and Determination of the Coke's Postreactive Strength CSR. 1. Thermodynamic Analysis of Coke Gasification in the Presence of H₂ and H₂O. Coke and Chemistry. December 1^o, 2023;66(12):603-12.

86. Li H, Chen J. An Analysis of Long-Process Ironmaking in a Reduction Smelting Furnace with Hydrogen-Enriched Conditions. Metals (Basel). October 1^o, 2023;13(10).

87. Henry S, Schneider N, Hughes S, Zoelle A, Grol E, Fout T, et al. Decarbonization of the Iron and Steel Sector: Challenges and Opportunities. SSRN Electronic Journal [Internet]. 2025; Available from: <https://www.ssrn.com/abstract=5070025>

88. Perpiñán J, Bailera M, Peña B. Full oxygen blast furnace steelmaking: From direct hydrogen injection to methanized BFG injection. Energy Convers Manag. November 1^o, 2023;295.

89. Huang JK, Li X, Hong SX, Gu TJ, Huang YB, Wang HR. DENSITY FUNCTIONAL THEORY (DFT) STUDY ON THE REACTION MECHANISM OF IN-SITU NO REDUCTION IN HYDROGEN RICH BLAST FURNACE. METALURGIJA. 2024;63:155-7.

90. Watari T, McLellan B. Global demand for green hydrogen-based steel: Insights from 28 scenarios. Int J Hydrogen Energy. 2024;79:630-5. doi:10.1016/j.ijhydene.2024.06.423.